

TABLE II

THIAMINE SYNTHESIS FROM OMP- AND TH-PHOSPHATE ESTERS WITHOUT ATP

The complete system contained 200 μ moles Tris buffer (pH 7.0), 10 μ moles cysteine, 0.1 μ mole OMP-P or OMP-PP, 0.1 μ mole Th, Th-P or Th-PP and 5 mg protein.

Substrates	Thiamine formed m μ mole 2 h
OMP-P + Th-P	1.4
OMP-P + Th-PP	0.4
OMP-P + Th	0
OMP-PP + Th	1.7
OMP-PP + Th-P	63.2
OMP-PP + Th-PP	22.2
OMP* + Th	11.3

* Original complete system containing 10 μ moles ATP and 10 μ moles $MgCl_2$.

thiamine formed¹ consists of free and phosphorylated thiamine and the enzyme system contains phosphatase activities.

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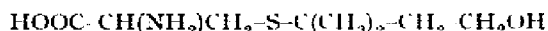
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On the biosynthesis of felinine

DATTA AND HARRIS¹ noted the existence of a new ninhydrin-positive spot upon paper chromatography of cat urine. WESTALL² isolated this material, which he called felinine, and obtained evidence indicating that its structure was that of S-(3-hydroxy-1:1-dimethylpropyl)cysteine



This structural assignment was confirmed by synthesis by TRIPPETT³.

We have developed an analytical method for felinine based on an ion-exchange chromatographic separation, followed by development of the ninhydrin color in the appropriate fractions, and have obtained evidence that cystine on the one hand, and either leucine or mevalonic acid on the other, can contribute to the formation

Abbreviation: MVA, mevalonic acid.

of this amino acid in the cat. The adult cat excretes more than 175 mg felinine/100 ml urine, whereas none of the other amino acids, with the exception of creatinine, are present at levels in excess of 6 mg/100 ml⁴. For this reason, application of a 0.2-ml aliquot of urine, purified by a slight modification of the procedure of WESTALL², to an Amberlite column in the acid form (0.9 × 20 cm, packed with Amberlite IR 120, Type III, particles 22-35 μ in diameter, conditioned and sized by the method of HAMILTON⁵) followed by elution with HCl (120 ml 0.37 *N* HCl, then 60 ml 1.1 *N* HCl, then 1.7 *N* HCl; collected 1-ml fractions, flow rate 30 ml/h) leads to the appearance of only 2 major ninhydrin-positive peaks. The first, which corresponds to felinine (established by comparing both on the column and by paper chromatography in two solvent systems with a sample of natural felinine) appears at approx. 140 ml, and the second, corresponding to creatinine, at 160-165 ml.

With the help of this analytical method, it was possible to show, as may be seen in Fig. 1, that the administration of cystine, leucine, or MVA to the cat markedly elevated urinary felinine excretion.

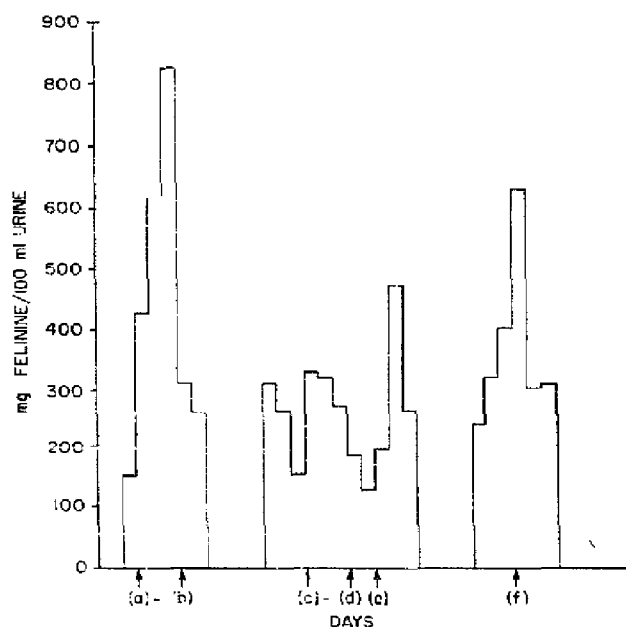


Fig. 1. Effect of administration of L-cystine, L-leucine and DL-MVA on the urinary excretion of felinine in the cat. (a)-(b): 1% L-cystine incorporated in the diet for 3 successive days starting at (a) and stopping at (b). (c)-(d): 1% L-leucine incorporated in the diet for 3 successive days starting at (c) and stopping at (d); at (e), a total of 1.5 g L-leucine dissolved in the minimum volume of phosphate buffer (0.1 *M*, pH 7.4) injected intraperitoneally in 3 doses of approx. 50 ml each. (f): Approx. 1.5 g DL-MVA (prepared from the DBED salt immediately prior to use) dissolved in phosphate buffer, injected intraperitoneally.

We have obtained confirmatory evidence for the participation of leucine and MVA in felinine biosynthesis by utilizing ¹⁴C-labeled compounds. Table I shows that significant activity is found in felinine when either DL-[2-¹⁴C]leucine or DL-[2-¹⁴C]-MVA are injected intraperitoneally into the cat. In the radioleucine experiment, felinine was the only compound in the urine containing significant radioactivity. With DL-[2-¹⁴C]MVA, however, by far the greater share of the activity in the urine was eluted from the ion-exchange column in the first fraction (0-20 ml). This material did not give a ninhydrin-reaction, was found to behave in the same way as a synthetic sample of DL-[2-¹⁴C]MVA upon paper chromatography and is assumed to be the unnatural form of [2-¹⁴C]MVA. That felinine is responsible for the radioactivity in the

felinine fraction from the ion-exchange column in both experiments, and not traces of highly active material, was also established by paper chromatography.

TABLE I
UTILIZATION OF DL-[2-¹⁴C]LEUCINE AND DL-[2-¹⁴C]MVA FOR FELININE
BIOSYNTHESIS IN THE CAT

Compound injected	Total activity (counts/min) in:	
	Radioactive precursor	Isolated compound
DL-[2- ¹⁴ C]Leucine*	779,000	Felinine 16,500
DL-[2- ¹⁴ C]Mevalonic acid**	970,000	MVA 408,000
		Felinine 5,100
		Creatinine*** 3,000

* 15 mg (specific activity, 0.05 mC/mmole) diluted to 200 mg with unlabeled DL-Leucine and recrystallized. Injected 100 mg.

** 14 mg (as DBED salt, specific activity 0.05 mC/mmole) converted to MVA and diluted to 140 mg with unlabeled DL-MVA. Injected 120 mg.

*** Identified by comparison of the elution position from the ion-exchange column, and the *R_F* value upon paper chromatography in *tert.* butanol-acetic acid-water, with the values for an authentic sample of creatinine.

Investigations bearing on the mechanism of biosynthesis of felinine, and upon the possibility that felinine biosynthesis plays a regulatory role in the metabolism of MVA or leucine, are under way in this laboratory.

It has recently come to our attention that Drs. SIMON BLACK and N. BAUMAN, National Institute of Arthritis and Metabolic Diseases, National Institutes of Health, Bethesda, Md., are also engaged in studying the biosynthesis of felinine. In unpublished work, they have found that [¹⁴C]acetate and DL-[2-¹⁴C]MVA can serve as precursors of radiofelinine.

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